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Variation of the Chemical Composition of
Water Emitted from the Prince of
Wales Feathers and Waikorohihi
Geysers During an Eruption of Each

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#### Abstract

In December, 1977 the authors collected 28 water samples from Prince of Wales Feathers Geyser and 17 water samples from Waikorohihi Geyser during an eruption of each and analyzed them.

As for Prince of Wales Feathers Geyser, the chloride and sulphate contents in the water were low at the start of eruption and increased with time. Thene, they showed instantaneously distinct low values when the discharge of Prince of Wales Feathers increased corresponding to the drop in water level in the neighbouring geyser Te Horu and also when the neighbouring geyser Pohutu started erupting. At the latter stage of the eruption they decreased gradually with time until its eruption stopped.

As for Waikorohihi Geyser, the chloride and sulphate contents in the water were low at the start of eruption. Soon after the start, they increased distinctly for a short time and then decreased gradually with time until the eruption stopped.

As the maximum and minimum values of chloride and sulphate contents, P. of W. F. showed 606 and 585 mg/l Cl, and 112 and 81 mg/l SO $_4$ ; Waikorohihi showed 615 and 568 mg/l Cl, and 101 and 83 mg/l SO $_4$ .

### I ) Introduction

On the North Island of New Zealand there were many geysers in the geothermal areas of the Taupo Volcanic Zone many years ago. However, some of them disappeared because of the construction of a dam for electric power or by boring to get more geothermal heat. To-day it is well known that in Whakarewarewa four geysers Prince of Wales Feathers, Pohutu, Waikorohihi and Mahanga Geysers are still active and emit hot water periodically. E. F.

<sup>\*</sup> Partly reported at the ACS/CSJ Chemical Congress held in Honolulu, Hawaii April 1-6, 1979.

Lloyd reported precisely the geology of Whakarewarewa Hot Springs in 1975<sup>1)</sup>. Aikawa and Noguchi visited New Zeal and to inspect geysers in December, 1977. A map of the thermal regions of New Zealand and a map of Whakarewarewa Hot Springs are shown in figures 1 and 2. On December 3, 1977 the water emitted from the orifice of Prince of Wales Feathers Geyser was collected in 28 polyethylene bottles during an eruption. Also, on December 4, 1977 the water emitted from the Waikorohihi Geyser was collected in 17 polyethylene bottles during an eruption. Both geysers are alkaline showing 8.9 and 8.7 in pH respectively.

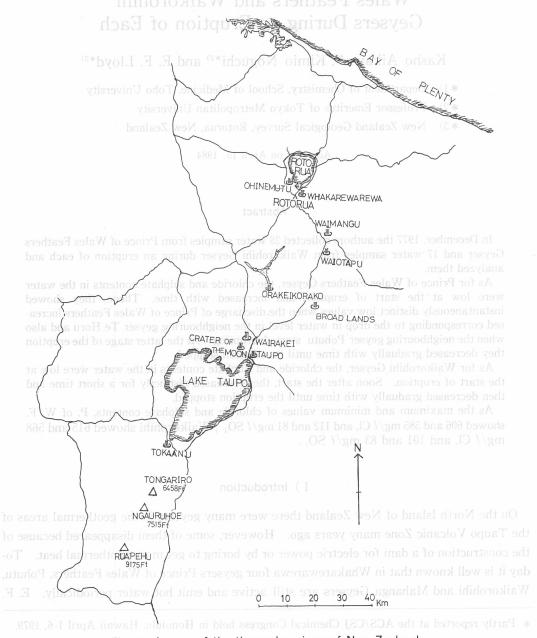
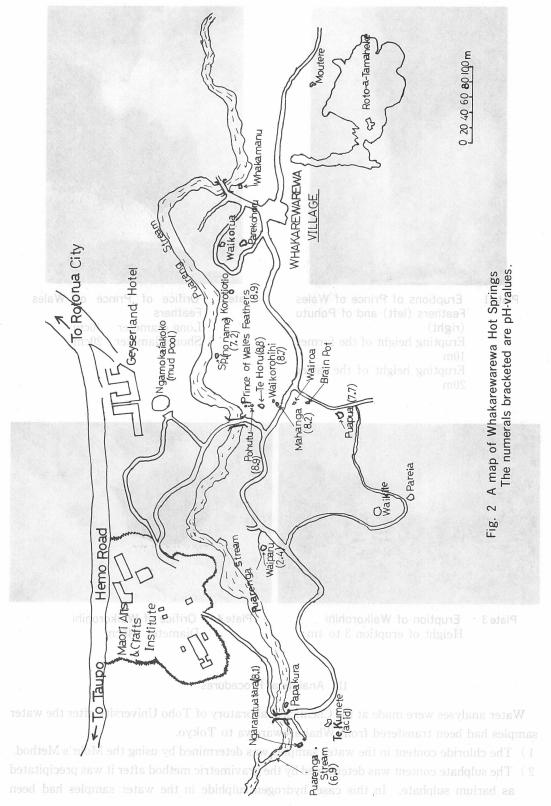


Fig. 1 A map of the thermal regions of New Zealand



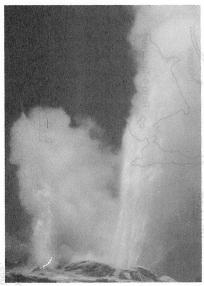


Plate 1 Eruptions of Prince of Wales
Feathers (left) and of Pohutu
(right)
Erupting height of the former
10m
Erupting height of the latter
20m



Plate 2 Orifice of Prince of Wales Feathers Long diameter 40cm Short diameter 20cm



Plate 3 Eruption of Waikorohihi Height of eruption 3 to 4m



Plate 4 Orifice of Waikorohihi
Diameter 40cm

# II) Analytical Procedures

Water analyses were made at the Chemical Laboratory of Toho University after the water samples had been transfered from Whakarewarewa to Tokyo.

- 1) The chloride content in the water samples was determined by using the Mohr's Method.
- 2) The sulphate content was determined by the gravimetric method after it was precipitated as barium sulphate. In this case, hydrogen sulphide in the water samples had been

oxidized completely to sulphate during transportation of the water samples from Whakarewarewa to Tokyo by ship.

tted from Prince of Wales Fea thers Geyser during its eruption

#### III) The Results

The analytical results of the water samples from Prince of Wales Feathers Geyser are shown in table 1 and fig. 3. As shown in fig. 3, the contents of chloride and sulphate in the water emitted from Prince of Wales Feathers were found to be low at the beginning of the eruption. Then, their contents increased with time. When its discharge increased distinctly corresponding to the drop in the water level in the neighbouring geyser Te Horu, the chloride and sulphate contents in water decreased instantaneously and then increased distinctly. A similar phenomenon occurred in its chloride and sulphate contents when the eruption of the neighbouring, great geyser Pohutu started erupting. After that, their contents decreased gradually with time. At the end of the long eruption which continued for 77.5 minutes, the water emitted from P. of W. F. was found to be low in chloride and sulphate contents as in the beginning stage.

It is presumed that the water emitted from the orifice of Prince of Wales Feathers and Waikorohihi Geysers is a mixture of the thermal water rich in chloride and sulphate and the shallow ground water poor in them. At the beginning stage of the eruption of P. of W. F. the water relatively rich in shallow ground water was emitted. Then, the water richer in thermal water was emitted increasing with time. When its discharge increased distinctly corresponding to the drop in the water level in the neighbouring geyser Te Horu, the water richer in shallow ground water was emitted instantaneously and 1.4 minutes later the water richer in thermal water was emitted again.

A similar phenomenon occurred in the water emitted from the orifice of P. of W. F. when the nearby Pohutu Geyser started erupting. After that, the water richer in ground water was emitted with time until the eruption stopped. Anyway, both at the beginning and ending stages of the eruption the water richer in shallow ground water was emitted.

As for Waikorohihi Geyser, the analytical results are shown in table 2 and fig. 4. In this case, at the start of the eruption the contents of chloride and sulphate were low, but soon after that, they showed distinctly higher values for short time and then decreased gradually with time. At the end of the eruption they showed slightly higher value again. The above mentioned results can be explained as follows: At the start of the eruption the water relatively rich in shallow ground water flowed out from the orifice of Waikorohihi and soon after that, the water rich in thermal water was emitted high up in the air for a short time. Then, with time the water richer in ground water was emitted. At the end of the eruption the water slightly rich in thermal water was emitted instantaneously high again and then, the eruption stopped immediately.

The maximum and minimum values of the chloride and sulphate contents and the rate of change are shown in table 3.

As for the cause of periodic eruption of the geysers, it is presumed that periodic eruptions of the geysers are caused by the relatively lower temperature water, namely, the water relatively

Whakarewarewa to Tokyo by ship teef contents in water emitted from Prince of Wales Feathers Geyser during its eruption

Observed on Dec. 3, 1977

The analytical results of the water samples from  $\Gamma_k^{0}$ OS e of 10 ales and samittees almas number 9 eruption started  $|mg/l| \approx (mg/l)$ en1) to 20.2 (min) at the w 590 d 90 erupting SI Ďπ. 10.0 Splashing to 2m 595 JL. Š 15.0 Waikorohihi starts erupting of V 2m n increased dist.71ctly (AA) 600 112 to 90.0 Discharge slightly increased 5) 27.2 597 104 30.6 Erupti JP. of drop 7) 602 93 30.9 Increase of discharge 120<sub>1</sub> 8) 31.8 from P. of 600 90 (mg/l)W. F. corre-110 32.8 sponding to drop in wa-10) 33.3 ter level in nearby Te Horu 11) 34.6 80 12) 600 610 591 13) 42.0 595 14) 48.1 Discharge of 600 P. of W. F. 590 increases CI 15) 48.4 580 597 16) 48.7 89 icher in ground 1911 ter (711s 1596) sw er 86 570 40 50 20 30 60 70 80 10 49.3 nearby Pohutu starts erupting (min) Fig. 3 Variation of the chloride and sulpha-50.0 590 19) te contents in the water emitted from  $wn_{000}$  table 2 and fi 90 50.2 Prince of Wales Feathers Geyser du-600 94 ring its eruption at the start of the 22) 603 Observed on Dec. 3, 1977 Discharge of Pohutu decreases in mentioned results can be explained as follows: At the start of the en<sub>thgish</sub>n the water relatively rich in shallow ground water flowed out from the (gnidsalge seniemmos) ohibi and soon 24) 56.7 65.3 585 66.0 Waikorohihi stops erupting 73.5 Pohutu stops erupting 587 26) 74.0 74.2 Discharge of P. of W.F. decreases suddenly 27) 74.6 585 As for the cause of periodic eruption of the geysers, 78 is pre-385 red the noitqurad 3.77 upt (82) of the geysers are caused by the relatively lower temperature water, namely beggotalter relatively

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Pohutu stopped erupting

Variation of the chloride and mixture of thermal water and steam. In this case, it retaw in the standlus boiling of the thermal water by pressure diddorokieWrilmonts bettime erupting) Geyser during its eruption Observed on Dec. 4, 1977 eruption eruption stopped erupting Sample Time since the C1 .⊆ \_ number eruption started (mg/l)(mg/l)erupting mes violent violent stopped (Pohutu starts (min) erupting 1) 0.0 Eruption 588 started of W.F. becomes Mahanga started 2) 0.2 615 101 3) 0.4 598 93 4) 0.7 589 92 Eruption nearby 5) 2.0 589 89 Erupti 6) 4.8 589 91 7) 9.7 585 90 (mg/1) 8) 13.3 582 88 100 9) 16.4 582 87 5042-20.5 nearby Mahanga stars erupting 10) 22.6 583 108 11) 31.9 583 88 620<sub>T</sub> 39.9 P. of W. F. becomes violent in eruption (erupts to 10m) 610-Pohutu starts erupting 600 12) 39.9 575 88 Cl 590 13) 46.2 575 87 Pohutu stops erupting 580 47.5 Pohutu stops erupting 570 P. of W. F. becomes violent in 52.8 eruption 560 54.0 Pohutu starts erupting 20 30 50 10 40 14) 55.3 570 85 Fig. 4 Variation of the chloride and sulpha-15) 59.4 568 83 te contents in the water emitted from 16) 59.8 570 88 Waikorohihi Geyser during its erup-17) 59.9 572 88 Eruption tion.

Table 3 The maximum and minimum values of chloride and sulphate contents and the rate of change

Observed on Dec. 4, 1977

Prince	of Wales	Feather	s Waiko	rohihi	
		C1	$SO_4$	Cl	$SO_4$
Maximum value	(mg/l)	606	112	615	101
Minimum value	(mg/l)	585	81	568	83
Difference	(mg/l)	21	31	47	18
Rate of change	(%)	3.5	27.7	7.6	17.8
Total number of		28	28	17	27
samples collected					

rich in ground water being heated in porous beds or in cavities under the ground with the mixture of thermal water and steam. In this case, it can be supposed that the steam was created by boiling of the thermal water by pressure decrease during its ascent from deep places.

## Acknowledgments

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### References

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